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Tricoccin R2

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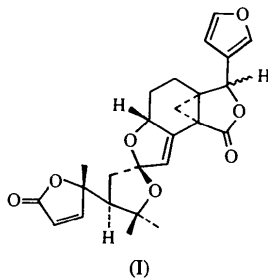
(Received 27 September 1994; accepted 24 May 1995)

Abstract

In the title compound, (3'*S*-{3'*α*,3'*αβ*,5'*αα*,7'*α*[*R**-(*R**)],8'*bβ*}-4-(2,5-dihydro-2-methyl-5-oxo-2-furanyl)-3'-(3-furanyl)-4,5,5',5'*α*-tetrahydro-5,5-dimethylspiro[furan-2(3*H*),7'(4'*H*)-1*H*,3*H*-3*α*,8*b*]methanobenzo[1,2-*b*:3,4-*c'*]difuran)-1'-one, C₂₅H₂₆O₇, some C—C bonds and C—C—C angles deviate significantly from their expected values. Both the terminal five-membered rings are nearly planar. The fused five-membered rings, *B*, *C* and *E*, have half-chair, distorted envelope and half-chair conformations, respectively. The six-membered ring is in a slightly distorted sofa conformation with a mean torsion angle of 33.5 (7)°. Puckering is pronounced near C11 and less so near C14. Packing of the molecules in the unit cell is stabilized by van der Waals interactions.

Comment

Tricoccin R2, (I) was isolated by Epe & Mondon (Herz, Grisebach, Kirby, 1983) from *Cneorum tricoccin* L., a shrub native to coastal areas of the western Mediterranean with hairless leaves, yellow blossoms and red fruits.



Some C—C bonds and C—C—C angles involved in the fused ring systems with axial substitutions of heavy bulky groups deviate by more than the 3σ level from their expected values (Engh & Huber, 1991). Two factors seem to be responsible for these deviations: (1)

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the degree of substitution at the C atom and (2) the presence of steric strain in the structure (Hall & Maslen, 1965; Gzella, Zaprutko, Wrezciono & Jaskólski, 1987; Sekar, Parthasarathy, Kundu & Barik, 1992, 1993). Rings *A* and *F* are planar within 2σ. Rings *B*, *C* and *E* are in half-chair, distorted envelope and half-chair conformations, respectively. Ring *D* is in a slightly distorted sofa conformation with a mean torsion angle of 33.5 (7)°. Puckering is pronounced near C11 and less so near C14. Packing of the molecules in the unit cell is stabilized by van der Waals interactions.

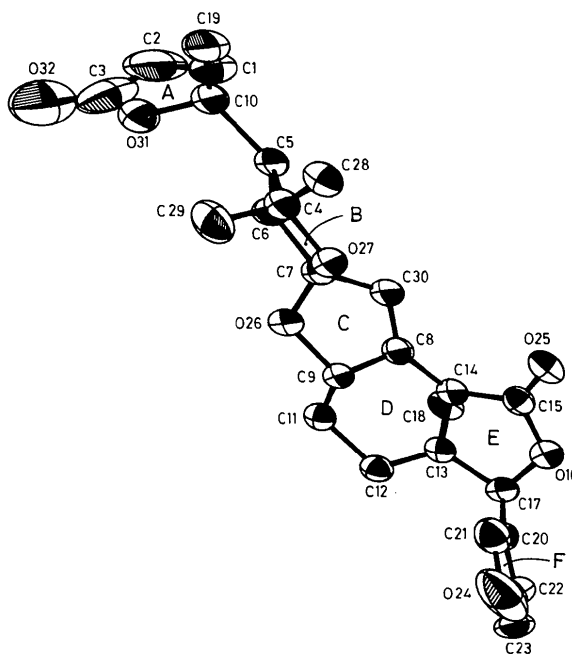


Fig. 1. Molecular structure with ellipsoids at the 50% probability level.

Experimental

Crystals were grown at room temperature from an ethanol/acetone mixture.

Crystal data

C₂₅H₂₆O₇
M_r = 438.48
 Monoclinic
*C*2
a = 22.939 (1) Å
b = 6.574 (2) Å
c = 16.481 (2) Å
 β = 114.67 (1)°
V = 2259 (2) Å³
Z = 4
D_x = 1.29 Mg m⁻³

Cu *Kα* radiation
 λ = 1.5418 Å
 Cell parameters from 25 reflections
 θ = 20–30°
 μ = 0.739 mm⁻¹
T = 293 K
 Needle
 0.30 × 0.25 × 0.20 mm
 Colourless

Data collection

Enraf-Nonius CAD-4
diffractometer
 θ - 2θ scans
Absorption correction:
empirical, ψ scan
 $T_{\min} = 0.961$, $T_{\max} =$
0.991
2312 measured reflections
2230 independent reflections

1996 observed reflections
[$I \geq 3.0\sigma(I)$]
 $R_{\text{int}} = 0.027$
 $\theta_{\max} = 70^\circ$
 $h = -26 \rightarrow 26$
 $k = 0 \rightarrow 8$
 $l = 0 \rightarrow 27$
2 standard reflections
frequency: 120 min
intensity decay: none

Refinement

Refinement on F
 $R = 0.049$
 $wR = 0.058$
 $S = 1.08$
1996 reflections
392 parameters
 $w = 1/[\sigma^2(F) + 0.004351F^2]$

$(\Delta/\sigma)_{\max} = 0.06$
 $\Delta\rho_{\max} = 0.298 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.424 \text{ e } \text{\AA}^{-3}$
Extinction correction: none
Atomic scattering factors
from *International Tables*
for *X-ray Crystallography*
(1974), Vol. IV, Table 2.2B)

Table 1. *Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)*

$$U_{\text{eq}} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i \cdot a_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	U_{eq}
C1	0.4996 (3)	0.3367 (13)	0.1535 (3)	0.086 (2)
C2	0.5369 (4)	0.3187 (24)	0.1109 (4)	0.132 (5)
C3	0.5549 (3)	0.1046 (29)	0.1167 (3)	0.152 (7)
C4	0.5110 (2)	-0.0331 (9)	0.3509 (2)	0.049 (1)
C5	0.5155 (2)	0.1504 (8)	0.2947 (2)	0.045 (1)
C6	0.5859 (2)	0.2098 (9)	0.3415 (2)	0.054 (2)
C7	0.6052 (2)	0.1500 (8)	0.4406 (2)	0.049 (1)
C8	0.6672 (2)	0.2698 (8)	0.5839 (2)	0.043 (1)
C9	0.6970 (2)	0.0756 (8)	0.5694 (2)	0.044 (1)
C10	0.4897 (2)	0.1402 (9)	0.1906 (2)	0.053 (2)
C11	0.7689 (2)	0.0942 (9)	0.6017 (2)	0.051 (1)
C12	0.7992 (2)	0.1337 (9)	0.7048 (2)	0.050 (1)
C13	0.7687 (2)	0.3112 (8)	0.7325 (2)	0.046 (1)
C14	0.7005 (2)	0.3743 (8)	0.6719 (2)	0.042 (1)
C15	0.6691 (2)	0.4274 (8)	0.7337 (2)	0.049 (2)
O16	0.7131 (1)	0.4224 (7)	0.8206 (2)	0.059 (1)
C17	0.7742 (2)	0.3347 (8)	0.8297 (2)	0.049 (1)
C18	0.7574 (2)	0.5110 (9)	0.6846 (3)	0.060 (1)
C19	0.4207 (2)	0.0713 (11)	0.1437 (3)	0.067 (2)
C20	0.7852 (2)	0.1452 (9)	0.8843 (2)	0.053 (2)
C21	0.7453 (2)	-0.0300 (10)	0.8696 (3)	0.068 (2)
C22	0.8365 (3)	0.1102 (11)	0.9621 (3)	0.076 (2)
C23	0.8346 (3)	-0.0704 (11)	0.9983 (3)	0.072 (2)
O24	0.7819 (5)	-0.1497 (13)	0.9468 (6)	0.133 (6)
O25	0.6141 (1)	0.4674 (8)	0.7145 (2)	0.066 (1)
O26	0.6653 (1)	0.0426 (7)	0.4734 (2)	0.055 (1)
O27	0.5551 (1)	0.0250	0.4412 (2)	0.060 (1)
C28	0.4469 (2)	-0.0559 (11)	0.3558 (3)	0.066 (2)
C29	0.5331 (4)	-0.2345 (11)	0.3272 (4)	0.084 (2)
C30	0.6163 (2)	0.3144 (9)	0.5104 (2)	0.051 (1)
O31	0.5280 (2)	-0.0026 (10)	0.1646 (2)	0.087 (2)
O32	0.5862 (2)	0.0030 (23)	0.0846 (3)	0.0226 (7)

Table 2. *Selected geometric parameters (\AA , $^\circ$)*

C4—C5	1.550 (7)	C11—C12	1.566 (5)
C5—C10	1.566 (5)	C13—C17	1.561 (5)
C6—C7	1.554 (5)		
C2—C3—O32	133.3 (9)	C14—C15—O25	128.3 (4)
C4—C5—C10	122.2 (4)	O16—C15—O25	121.3 (5)
C14—C8—C30	133.5 (5)	C17—C20—C22	125.1 (6)
C12—C13—C17	122.2 (4)	C17—C20—C21	129.4 (5)
C8—C14—C15	124.4 (5)		

Ring B

C4—C5—C6—C7 -30.3 (6)
C5—C6—C7—O27 13.4 (6)
C6—C7—O27—C4 10.6 (6)
C7—O27—C4—C5 -29.6 (6)
O27—C4—C5—C6 36.3 (6)

Ring C

C7—C30—C8—C9 2.4 (7)
C30—C8—C9—O26 12.3 (7)
C8—C9—O26—C7 -22.7 (6)
C9—O26—C7—C30 23.8 (6)
O26—C7—C30—C8 -16.4 (7)

Ring D

C8—C9—C11—C12 62.5 (6)
C9—C11—C12—C13 -52.3 (7)
C11—C12—C13—C14 24.0 (8)
C12—C13—C14—C8 -4.3 (8)
C13—C14—C8—C9 13.9 (8)
C14—C8—C9—C11 -44.1 (7)

Ring E

C13—C14—C15—O16 6.4 (7)
C14—C15—O16—C17 -9.7 (7)
C15—O16—C17—C13 8.8 (7)
O16—C17—C13—C14 -4.3 (6)
C17—C13—C14—C15 -0.9 (6)

Refinement was by the full-matrix least-squares method. H atoms were refined isotropically taking the starting values of the displacement parameters to be those of the heavy atoms to which they are covalently bonded.

Data collection: *CAD-4 Software* (Enraf-Nonius, 1989). Program used to determine the space group: *STATCW* (Sekar, 1991). Program used to solve structure: *SHELXS86* (Sheldrick, 1990). Program used to refine structure: *SHELXL76* (Sheldrick, 1976). Program used to calculate the molecular parameters: *PARST* (Nardelli, 1983). Molecular graphics: *ORTEPII* (Johnson, 1976).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: VJ1017). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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5-Methyl-3-(1-phenylethylidene)-2,3-dihydrobenzo[*b*]furan-2-one

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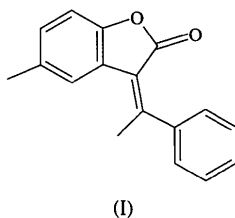
(Received 12 December 1994; accepted 24 July 1995)

Abstract

The benzofuran ring in the title compound, C₁₇H₁₄O₂, is planar with a methyl group substituted in the 5-position. The 5-methyl and 3-ethylidene groups are in *trans* positions with respect to the benzofuran ring. The structure is stabilized by intermolecular van der Waals interactions.

Comment

The title compound, (I), was prepared as part of a study of diastereoselectivities of free-radical reactions. The crystal structure was determined in order to establish whether a phenylethylidene group would be formed *trans* or *gauche* to the benzofuran ring.



The benzofuran ring is planar (with a maximum deviation from the plane of 0.75 Å for C8) and shows typical aromaticity and delocalization of π electrons. The phenyl ring is twisted out of the plane of the benzofuran moiety by 130°. The 5-methyl and ethylidene groups are in *trans* positions with respect to the benzofuran ring. Bond lengths and bond angles are as expected.

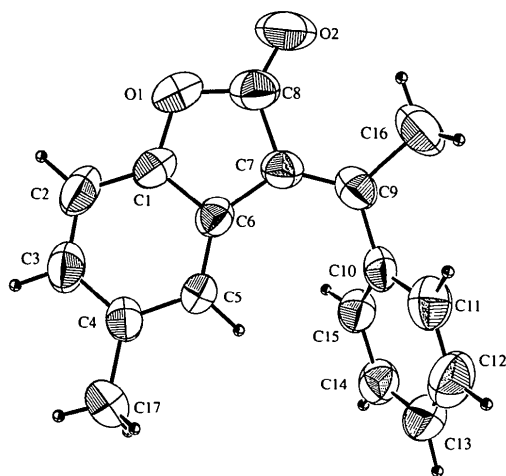


Fig. 1. View of the title molecule with displacement ellipsoids plotted at the 50% probability level.

Experimental

Crystal supplied by S. Selladurai.

Crystal data

C₁₇H₁₄O₂

$M_r = 250.3$

Triclinic

$P\bar{1}$

$a = 9.890$ (5) Å

$b = 9.258$ (4) Å

$c = 8.878$ (5) Å

$\alpha = 108.26$ (4)°

$\beta = 78.48$ (4)°

$\gamma = 67.38$ (4)°

$V = 663.3$ (6) Å³

$Z = 2$

$D_x = 1.253$ Mg m⁻³

$D_m = 1.249$ Mg m⁻³

D_m measured by flotation

Data collection

Syntax P3 diffractometer

$\theta/2\theta$ scans

Absorption correction:

none

1901 measured reflections

1705 independent reflections

913 observed reflections

$[I > 3\sigma(I)]$

Mo K α radiation

$\lambda = 0.71073$ Å

Cell parameters from 25 reflections

$\theta = 11.5$ – 15°

$\mu = 0.0757$ mm⁻¹

$T = 293$ K

Block

$0.35 \times 0.29 \times 0.12$ mm

Red

$\theta_{\max} = 25^\circ$

$h = -11 \rightarrow 11$

$k = 0 \rightarrow 10$

$l = -10 \rightarrow 10$

3 standard reflections

monitored every 97

reflections

intensity decay: <5%

Refinement

Refinement on F

$R = 0.063$

$wR = 0.083$

$S = 1.271$

913 reflections

214 parameters

H-atom parameters not

refined

$w = 1/\sigma^2(F)$

$(\Delta/\sigma)_{\max} = 0.58$

$\Delta\rho_{\max} = 2.2$ e Å⁻³

$\Delta\rho_{\min} = 0.6$ e Å⁻³

Extinction correction: none

Atomic scattering factors

from Cromer & Mann

(1968)